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5 **Materials and Methods**

6 Biochars were produced using BEST Energies continuous slow pyrolysis process. Chicken litter
7 (CL) was pyrolysed with a highest temperature of treatment (HTT) of 450°C and held at this
8 temperature for 30 min. Greenwaste (GW) and paper mill sludge (PM) were both pyrolysed with
9 a HTT of 550°C and residence time of 60 min. The CL feedstock also included sawdust, whereas
10 the PM feedstock had woodchip as secondary constituent. Therefore, all feedstocks had 20–30%
11 of woody material. The three fresh biochars differed in some key chemical characteristics in that
12 (i) the PM biochar had a very high acid neutralizing capacity (29% CaCO₃ equivalence
13 compared with 14% for the CL biochar and 0.5% for the GW biochar), and (ii) the CL biochar
14 had a high P and N content compared with the other two biochars. These biochars were
15 incorporated in a Ferrosol in a sub-tropical environment at the Wollongbar Research Station
16 (Industry and Investment NSW, Australia; 28°50'S, 153°25'E)). A summer sweet corn, winter
17 pulse rotation was established at the site.

18 Soil samples (*ca.* 1 kg) were taken after the first and the second year of biochar incorporation
19 from three sites at each experimental plot to a depth of 10 cm and then pooled. The fresh wet soil
20 was dispersed by agitation in ten volumes of demineralised water (1:10). The soil/biochar mix
21 was sieved using a 750 µm sieve. Biochar pieces were manually removed from the sieve surface
22 using a jeweler forceps. Light biochar pieces were recovered from the supernatant by
23 centrifugation at 2000 x G. Samples were then washed 3 times with demineralised water in a

24 beaker with slight agitation from a magnetic stirrer, dried and then stored dry in an airtight
25 container at 4°C.

26 Biochar samples were mounted on aluminum stubs and carbon coated prior to the analysis using
27 scanning electron microscopy (SEM), energy X-ray dispersive spectrometry (EDS) and
28 transmission electron microscopy (TEM). Samples were examined in a Hitachi 3400 SEM
29 (Japan) to which EDS facilities were attached. Electron transparent sections for TEM specimens
30 were prepared using a focused ion beam (FIB) (FEI Nova Nanolab 200 dual beam FIB). More
31 details of the sectioning technique are found in Giannuzzi (2005). All TEM specimens were
32 examined using a Philips CM200 field emission gun TEM. Scanning TEM mode (STEM) was
33 used to perform EDS mapping and line scans (individual points were taken at 1 nm intervals) in
34 the region of interest.

35 The atomic concentrations of elements at the surface of biochars, as well as their chemical
36 bonding, were investigated by X-ray photoelectron spectroscopy (XPS) (ESCALAB-220i-XL,
37 VG Scientific, UK) with a monochromatic Al K α X-ray source ($h\nu = 1486.6\text{eV}$) induced by 10
38 kV, 15 mA Al K α radiation with a spot size approximately 1 mm in diameter. The pass energy
39 for survey scans was 100eV with a scan step of 1eV, and for region scans, 20eV with a scan step
40 of 0.1eV was used. Neutralization was applied by using a 4eV electron flood source to avoid
41 surface charging. A peak shift was performed using the C1s peak at 285eV as a reference.
42 Photoelectrons were collected at a take-off angle of 90°. XPS spectra were analyzed and
43 deconvoluted using the Gaussian-Lorentzian sum function with a different % Gaussian-
44 Lorentzian value to optimize the spectra. A Shirley background correction was carried out to
45 remove background noise. For non-metal elements, the quantitative measurement error is around
46 1–5%, but can be minimized to 0.5–1% by increasing region scan times.

47 Solid state ^{13}C magic angle spinning (MAS) NMR spectra were obtained at a frequency of 50.3
48 MHz using a Varian (Palo Alto, CA) Unity200 spectrometer. Samples were packed in a 7 mm
49 diameter cylindrical zirconia rotor with Kel-F end caps and spun at the “magic angle” (54.7) at
50 5000 ± 100 Hz in a Doty Scientific (Columbia, SC) MAS probe. Free induction decays (FIDs)
51 were acquired with a sweep width of 40 kHz; 1216 data points were collected over an acquisition
52 time of 15 ms. All spectra were zero-filled to 8192 data points and processed with a 50 Hz
53 Lorentzian line broadening and a 0.010 s Gaussian broadening. Chemical shifts were externally
54 referenced to the methyl resonance of hexamethylbenzene at 17.36 ppm. Cross polarisation (CP)
55 spectra represent the accumulation of 4000 scans and were acquired using a 90° pulse of 5–6
56 μs duration, a 1 ms contact time and a 1 s recycle delay. Direct polarisation (DP) spectra
57 represent the accumulation of 500–1000 scans and were acquired using a 90° ^{13}C pulse of 5–10
58 μs duration and a 90 s recycle delay. Spin counting was carried out using the method of Smernik
59 and Oades (2000). Briefly, spin counting experiments involve comparison of the ^{13}C NMR
60 integrated signal intensity of a sample with that of a standard with a known mass and carbon
61 content. Glycine (AR grade, Ajax Chemicals) was used as external reference material. Aliquots
62 of each heat-treated material (200 mg) were placed in 2 ml glass vials with PTFE-lined screw
63 caps. Neat ^{13}C -labelled sorbates (ca. 10 μl) were added using a micropipette and the samples
64 were shaken for ca. 1 min and stored at room temperature until analysis using NMR.

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66 Giannuzzi LA, Stevie FA (2005) ‘Introduction to Focused Ion Beams Instrumentation, Theory,
67 Techniques and Practice’, Springer, New York.

68 Smernik RJ, Oades JM (2000) The use of spin counting for determining quantitation in solid
69 state ^{13}C NMR spectra of natural organic matter 1. Model systems and the effects of
70 paramagnetic impurities. *Geoderma* **96**, 101–129.