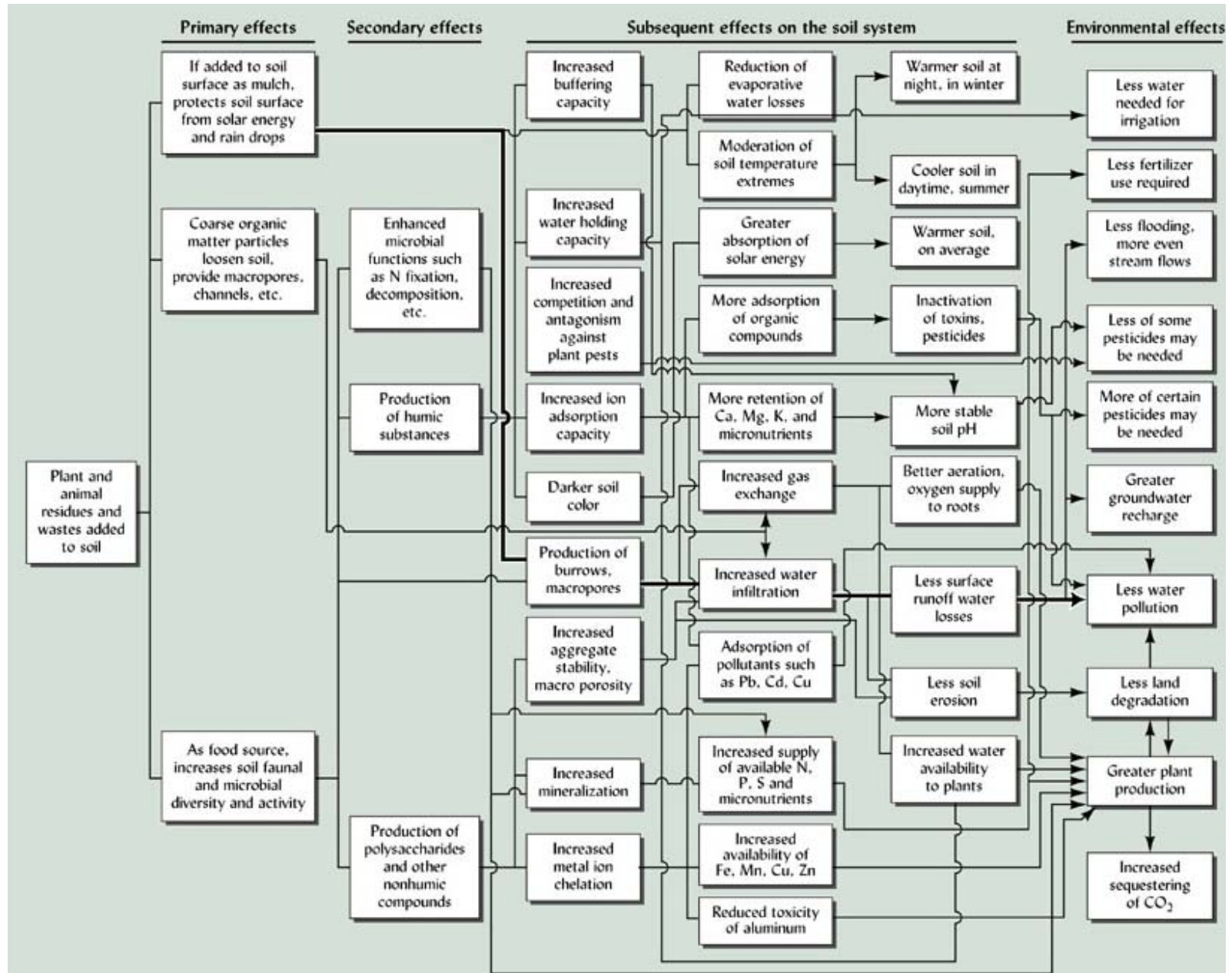
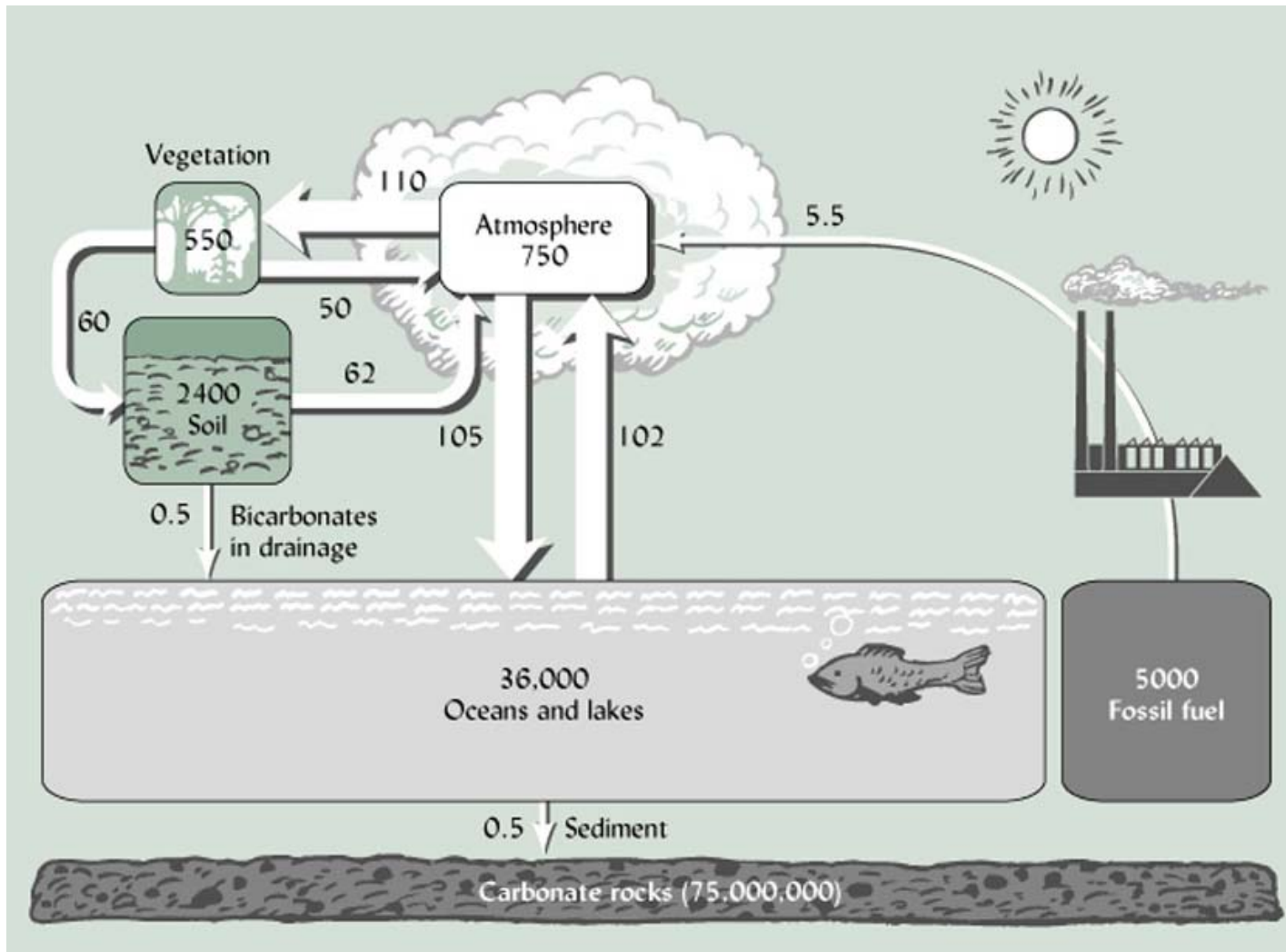


# Soil Organic Matter

Figure 12.15



# Global Carbon Cycle

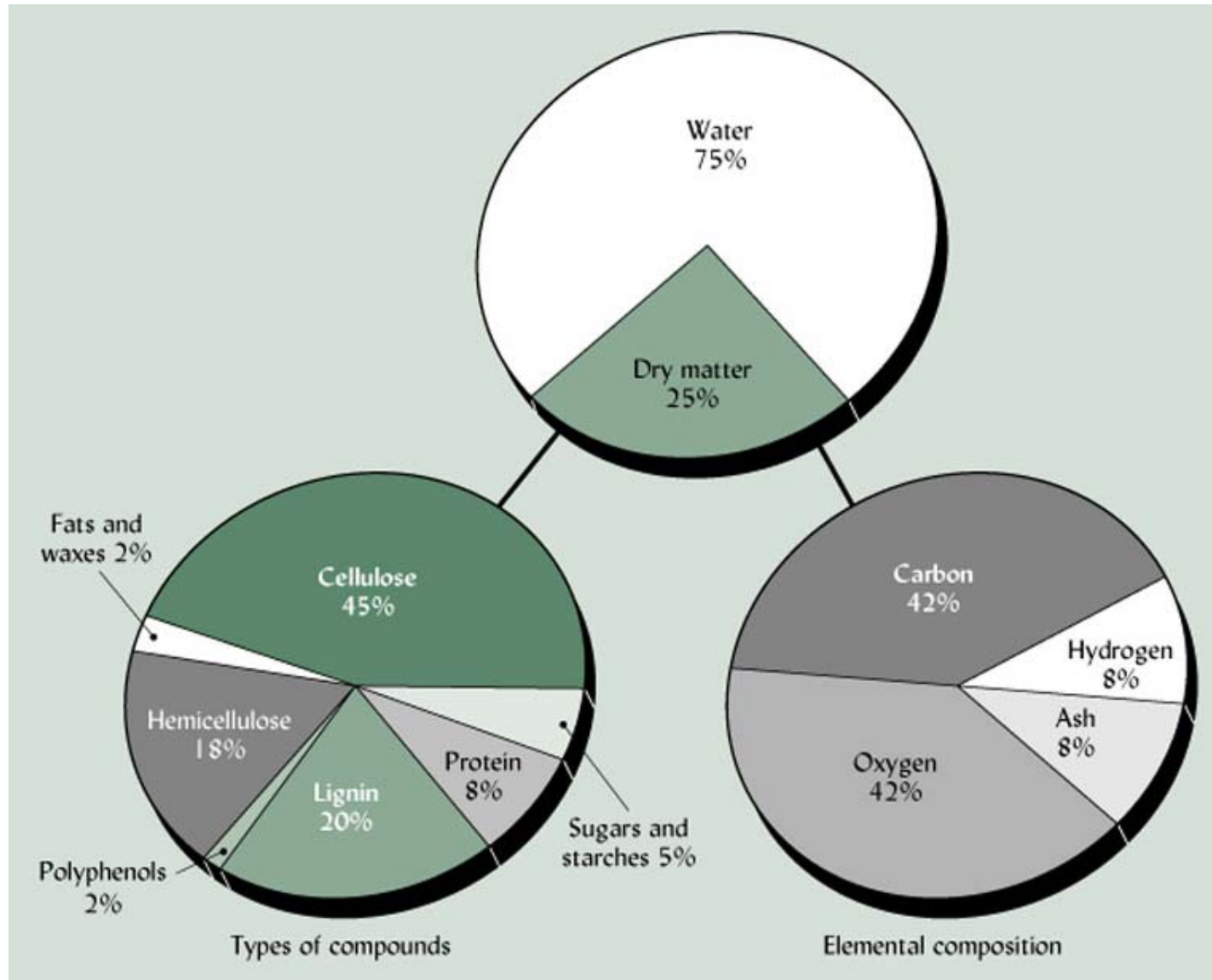


(1) More C is entering the atmosphere than is being absorbed by the oceans and (2) more C is leaving the soil than is entering

What is causing this and what might be the consequence of this?

Petagrams = Pg =  $10^{15}$  g

## Organic Matter – what is it?



Why is this important – this is the source of soil C

## **SOIL ORGANIC MATTER**

**Mineral soils generally have 2-5% organic matter (by weight).**

**Organic matter profoundly affects soil properties.**

**Elemental composition: mostly carbon (C)  
lesser amounts of  
H,O,P,N,S**

**Main source of N & S for plants in unfertilized soils.**

**Forms of Organic Matter:**

- living & dead plant material**
- soil microorganisms**
- humus -> derived from plant decay**
  - > complex colloidal molecules**
  - > no visible features of plant tissue**



**TABLE 12.1 Mass of Organic Carbon in the World's Soils**

*Values for the upper 1 m represent most of the carbon in the soil profile. The upper 15 cm generally represents the surface soil, which is most readily influenced by land use and soil management.*

Soil order	Global area, $10^3 \text{ km}^2$	Organic carbon <sup>a</sup> in upper 100 cm			Organic carbon <sup>a</sup> in upper 15 cm	
		Mg/ha	Global Pg <sup>b</sup>	% of global	Range, <sup>c</sup> %	Typical, <sup>c</sup> %
Entisols	14,921	99	148	9	0.06–6.0	— <sup>d</sup>
Inceptisols <sup>e</sup>	21,580	163	352	22	0.06–6.0	— <sup>d</sup>
Histosols <sup>e</sup>	1,745	2,045	357	23	12–57	47
Andisols	2,552	306	78	5	1.2–10	6
Vertisols	3,287	58	19	1	0.5–1.8	0.9
Aridisols	31,743	35	110	7	0.1–1.0	0.6
Mollisols	5,480	131	73	5	0.9–4.0	2.4
Spodosols	4,878	146	71	5	1.5–5.0	2.0
Alfisols	18,283	69	127	8	0.5–3.8	1.4
Ultisols	11,330	93	105	7	0.9–3.3	1.4
Oxisols	11,772	101	119	8	0.9–3.0	2.0
Misc. land	7,644	24	18	1	—	—
Total	135,215		1576	100		

<sup>a</sup> Organic matter may be roughly estimated as 1.7 to 2.0 times this value. The value traditionally used is 1.72. Organic nitrogen may also be estimated from organic carbon values by dividing by 12 for most soils, but see Section 12.3.

<sup>b</sup> Petagram =  $10^{15}$  g.

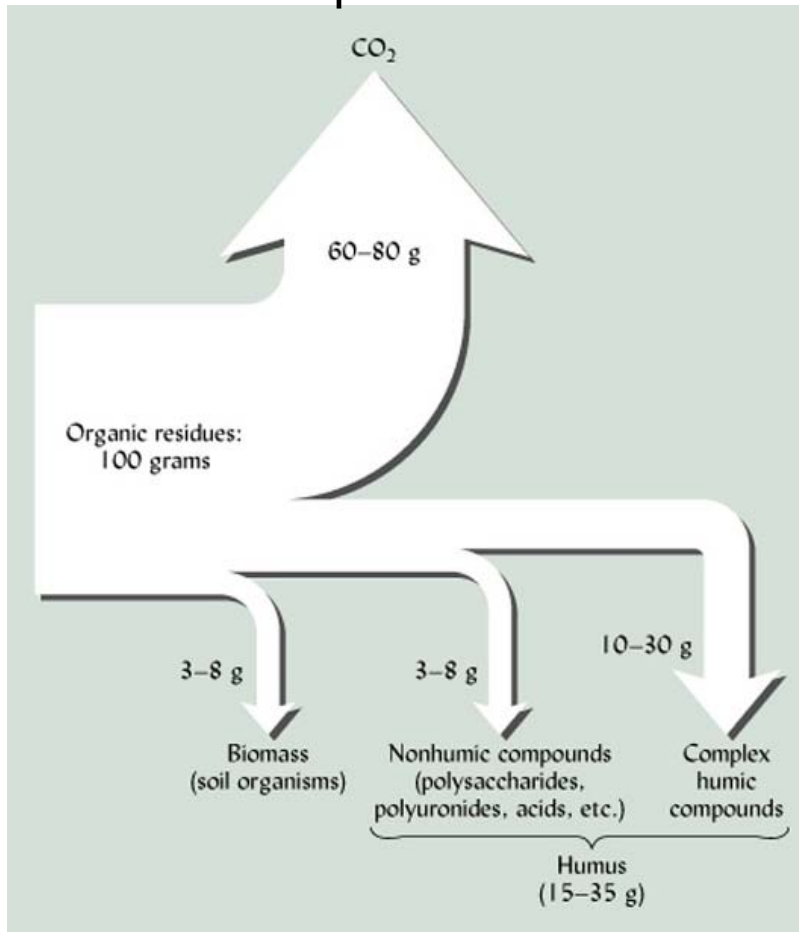
<sup>c</sup> Percent on mass basis (i.e., g/100 g).

<sup>d</sup> These soils are too variable to suggest a typical value.

<sup>e</sup> Carbon stored in Gelisols is included with these soils.

Data calculated from Eswaran, et al. (1993) and Brady (1990).

We generally perceive OM decomposition as...



1. **CONSUMES OXYGEN (O<sub>2</sub>)**
2. **PRODUCES CARBON DIOXIDE (CO<sub>2</sub>)**

and while the basic building blocks of all organic carbon structures are the same (C, H & O)

(1) not all organic carbon structures are the same

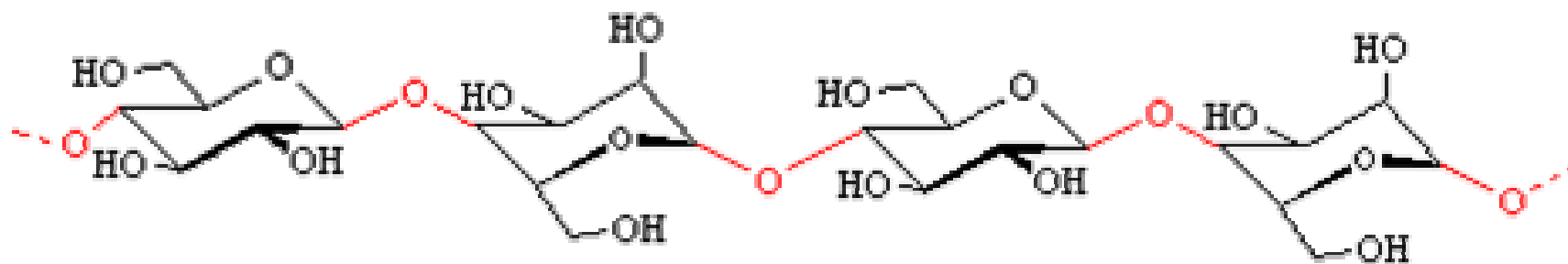
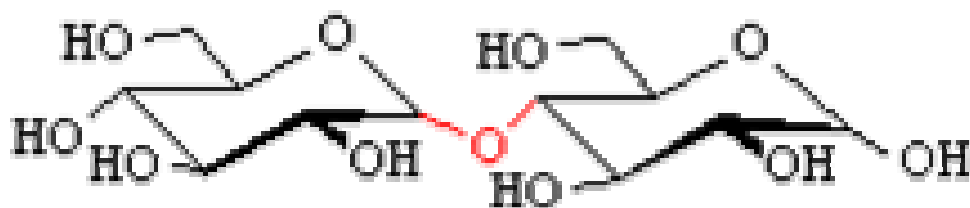
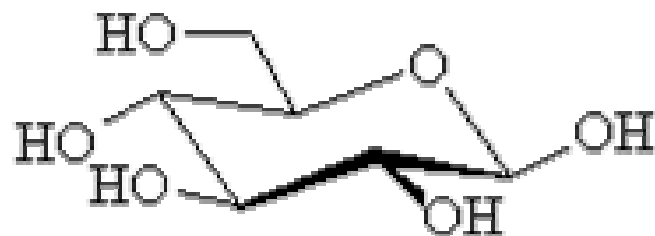
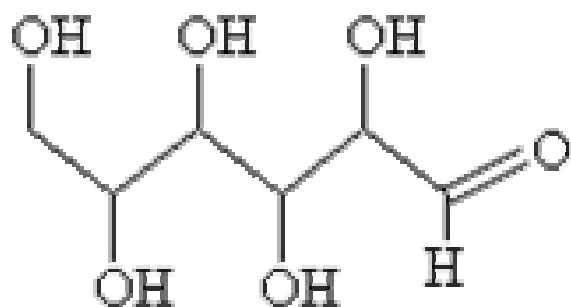
**and**

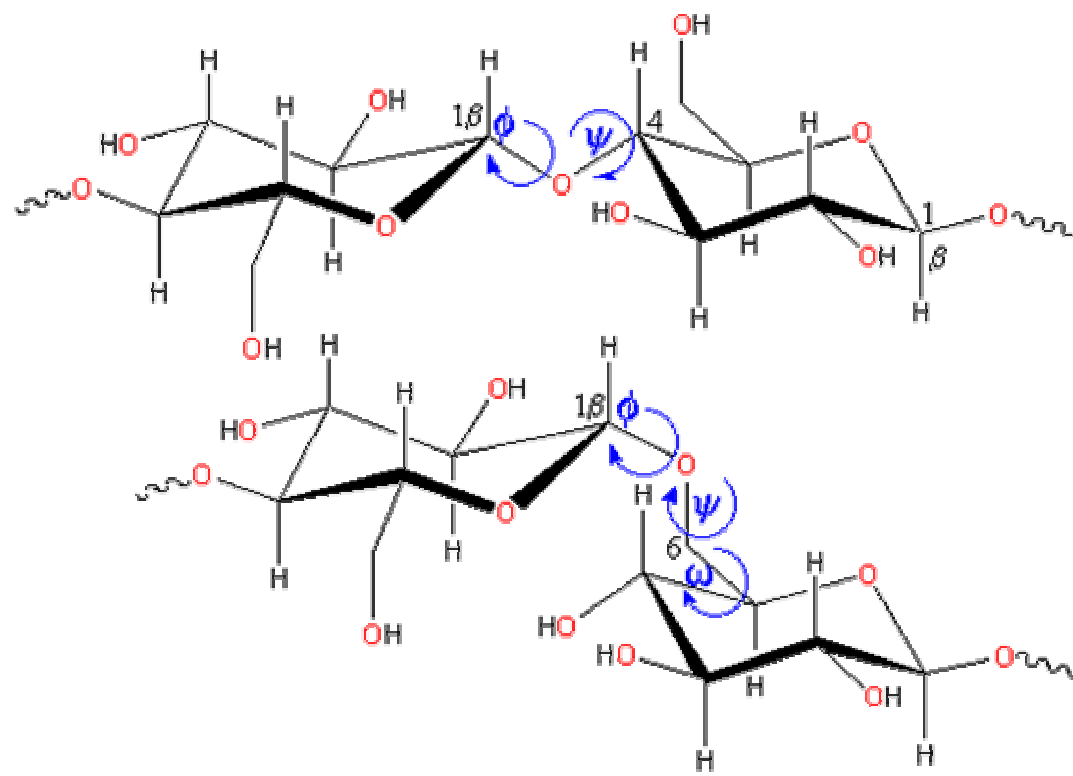
(2) not all are easy to break down

Terminology watch ~

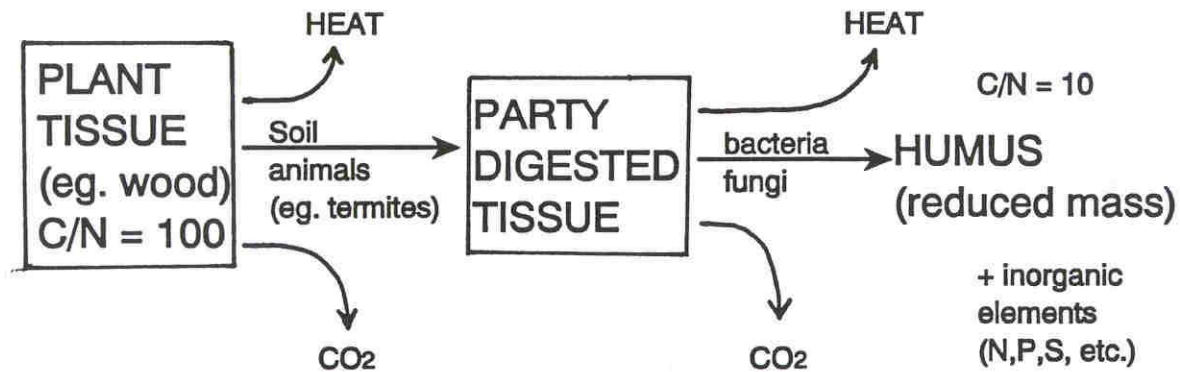
Mineralization = the conversion of an element from an organic state (biomass, etc.) to an inorganic state

Immobilization = the conversion of an element from an inorganic state to an organic state (biomass)





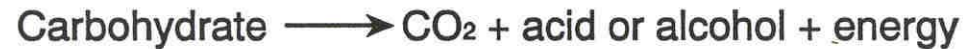




Respiration (aerobic):



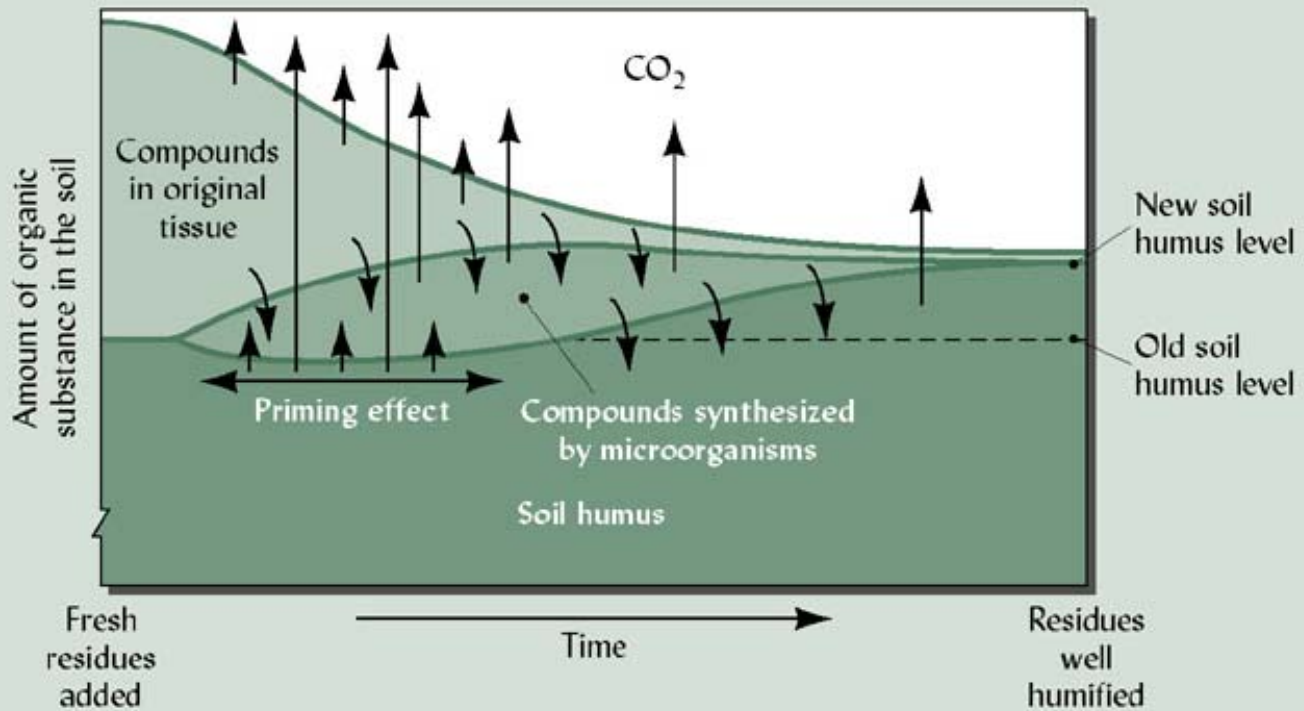
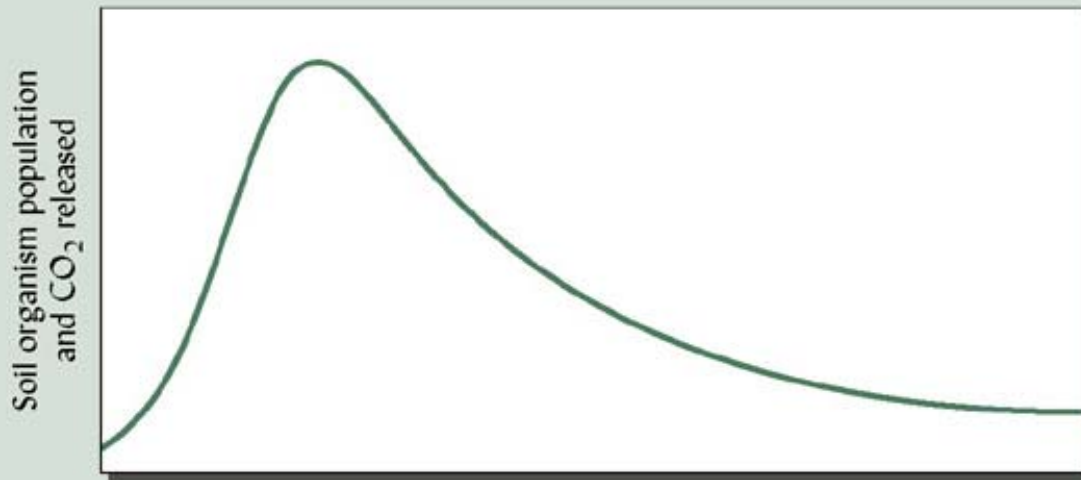
Fermentation (anaerobic):



Rate of CO<sub>2</sub> release = measure of microbial activity

Effects of Respiration on Soil:

1. Raises CO<sub>2</sub> in soil air by 10-1000x
2. Lowers soil pH (carbonic acid)
3. Lowers O<sub>2</sub> level; potential for anoxia



## Recognizable compounds-

Polysaccharides

rapid  
rate of decomposition  
very slow

**Glucose**

- simple sugar

**Starch**

- building block for cellulose

**Protein**

- water soluble and crude protein

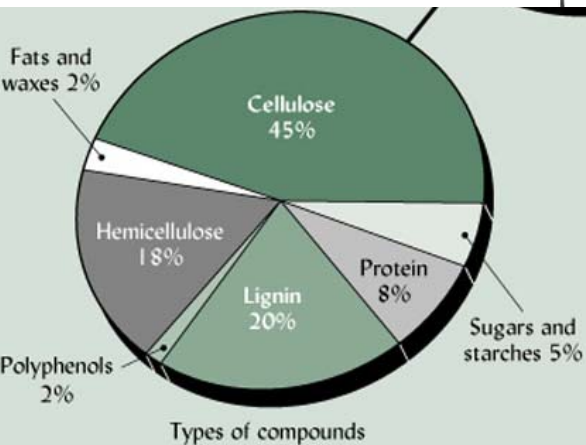
**Cellulose & Hemicellulose**

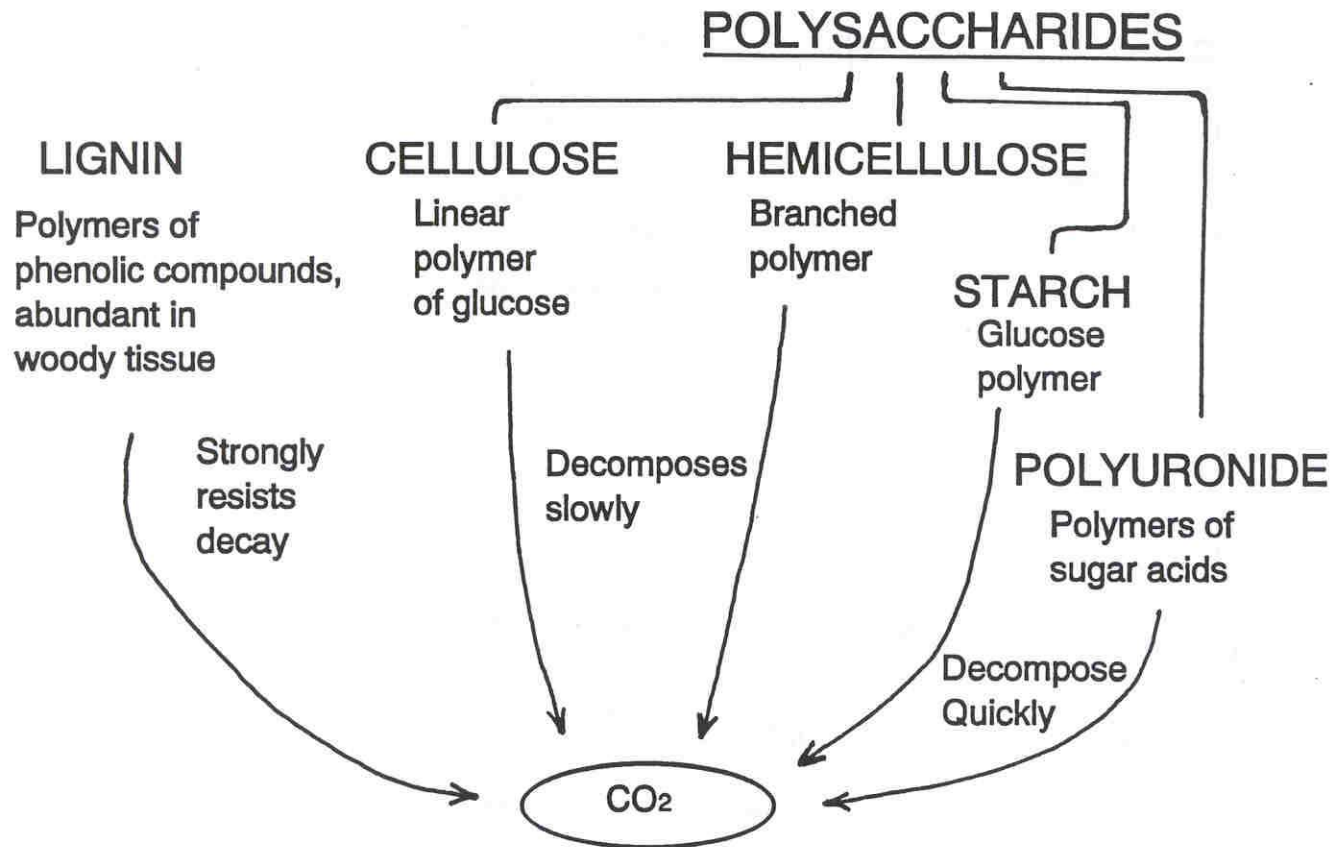
- most abundant constituent of plant tissue
- source of food for soil microbes

**Fats, waxes, and tannins-**

**Lignin**

- major constituent of plant tissue
- polymers (very large molecules) help to aggregate mineral grains
- slowly decompose, part of humus

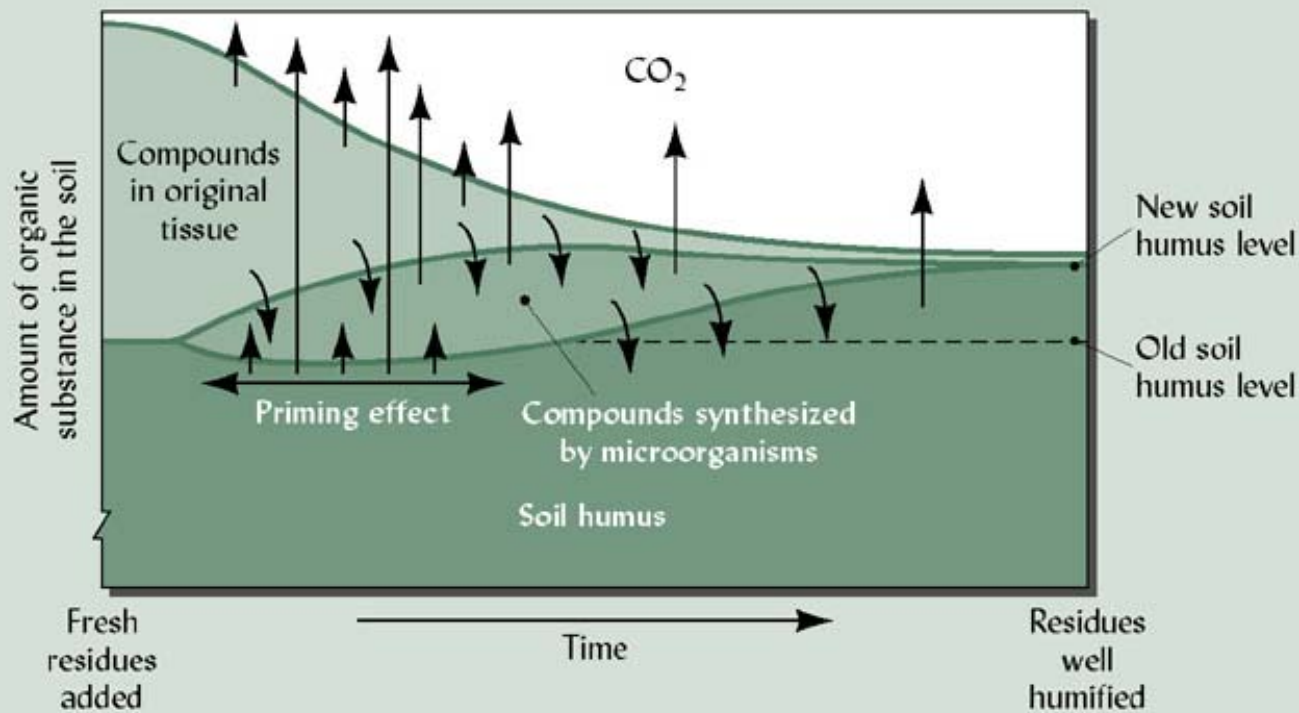
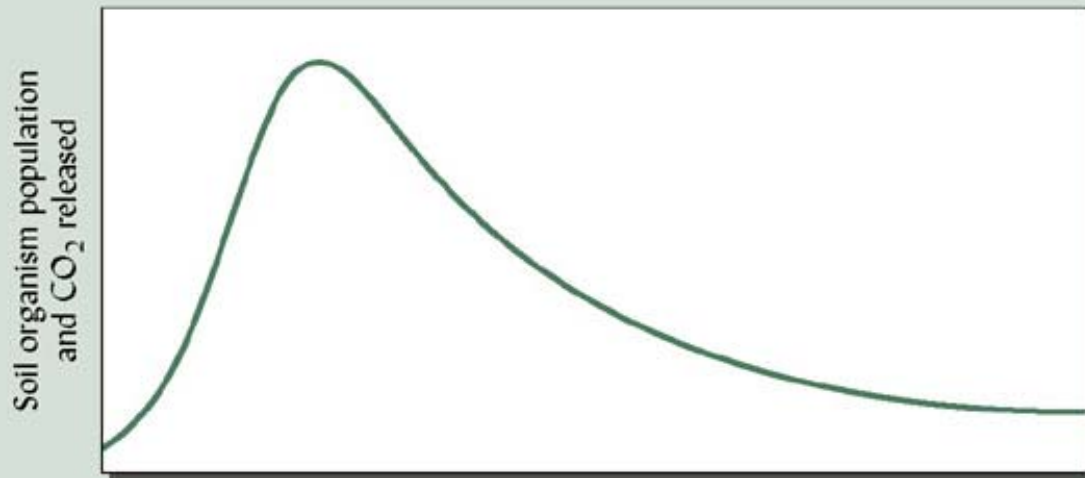




Result: Lignin-like compounds accumulate in humus because lignin is preserved.

Resistance of Lignin to decay because:

1. Microbes cannot absorb the polymer, must release EXTRACELLULAR ENZYMES (only specialized FUNGI & BACTERIA)
2. Woody tissue (lignin) is very low in nitrogen, inhibits biological activity.
3. Low exposed surface area.





# Factors controlling the rate of OM decay

- 1 Placement
- 2 Size and Surface Area
- 3 C:N Ratio
- 4 Litter Quality

## 1. Placement

- surface placement vs. incorporation into the soil

heating and cooling as well as wetting and drying

vs.

constant climatic conditions and intimate contact with soil organisms

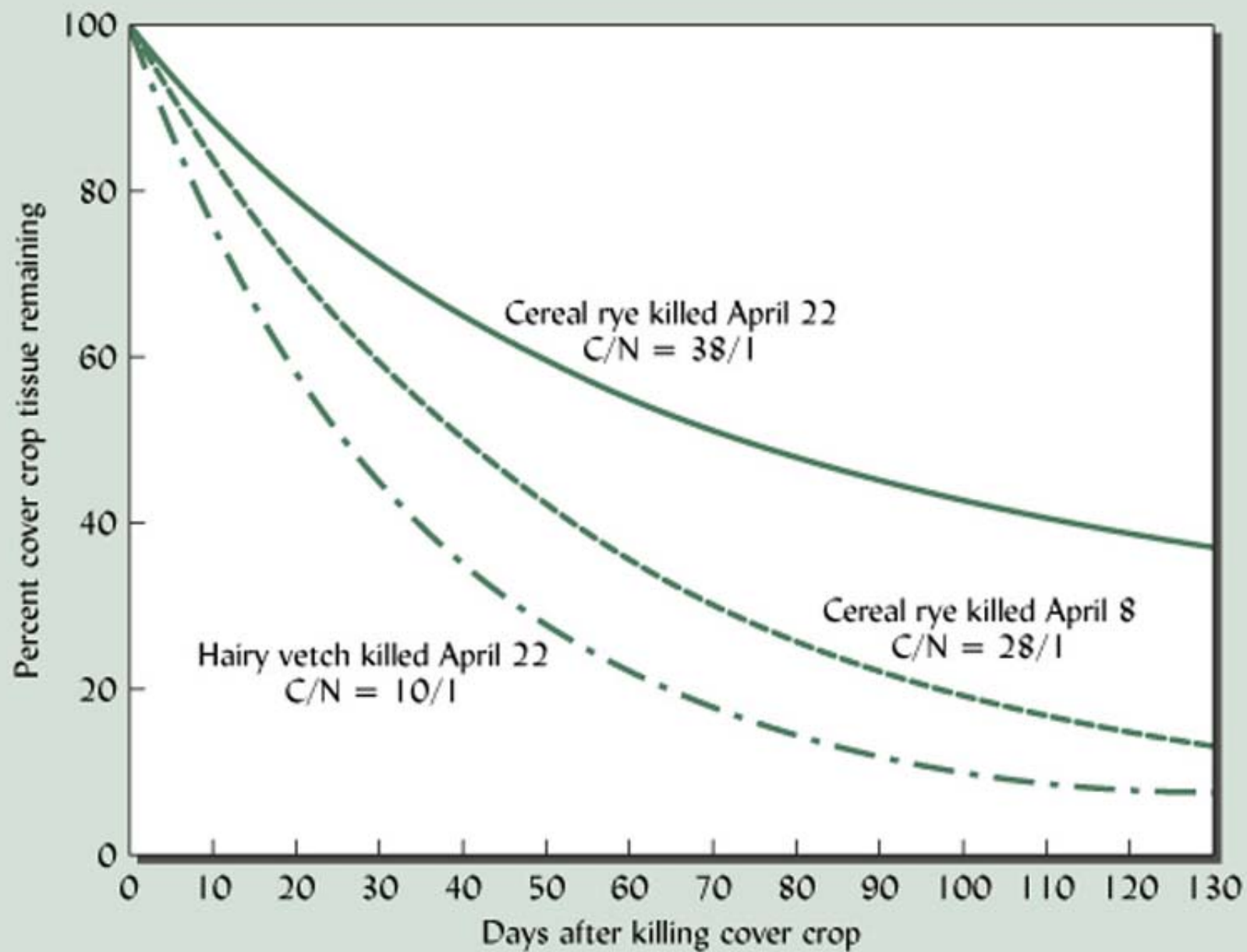
## 2. Size and Surface Area

- smaller vs. larger particles

Surface area increases with smaller particles, therefore more contact areas for organism doing the decomposition

## 3. C:N Ratio

- intense competition for N & as such the ratio determines the rate of decay

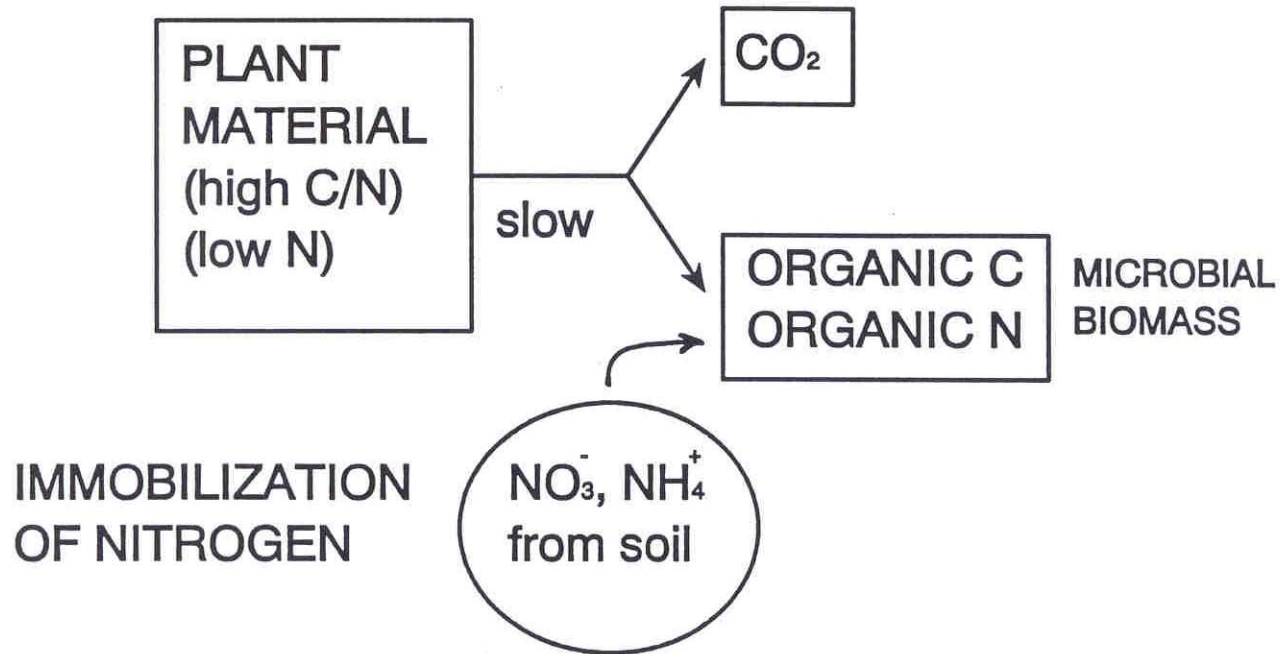


<b><u>Typical C/N Ratios</u></b>	<b><u>C/N</u></b>
<b>Tree leaves</b>	<b>45-70</b>
<b>Tree roots</b>	<b>60-90</b>
<b>Wood (oak)</b>	<b>130-400</b>
<b>Clover leaves</b>	<b>8-16</b>
<b>Grass shoots (young)</b>	<b>12-15</b>
<b>Bacteria</b>	<b>5-14</b>

Look at Table 12.2 for a more complete listing!

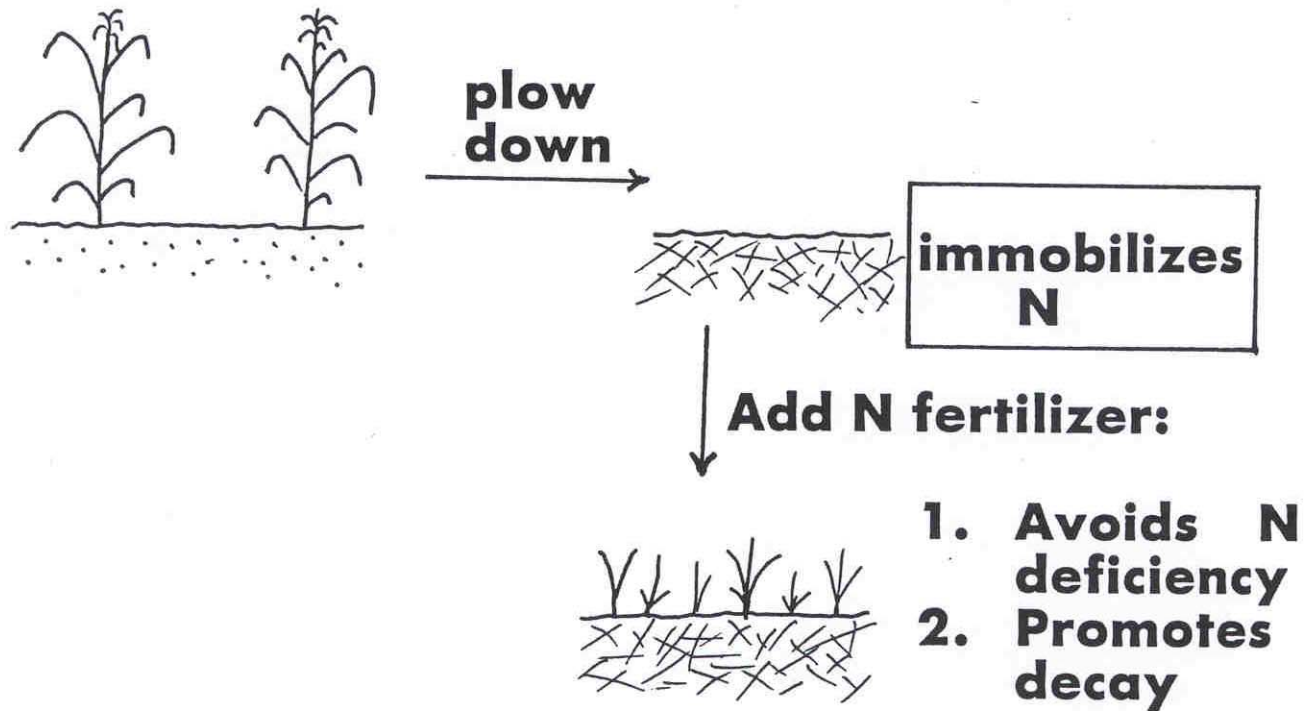
## Nutrient Release and Immobilization

$C/N > 20$  (% N < 2.5)  $\rightarrow$  N immobilized



## **Example:**

### **Corn Stover (high C/N ratio):**



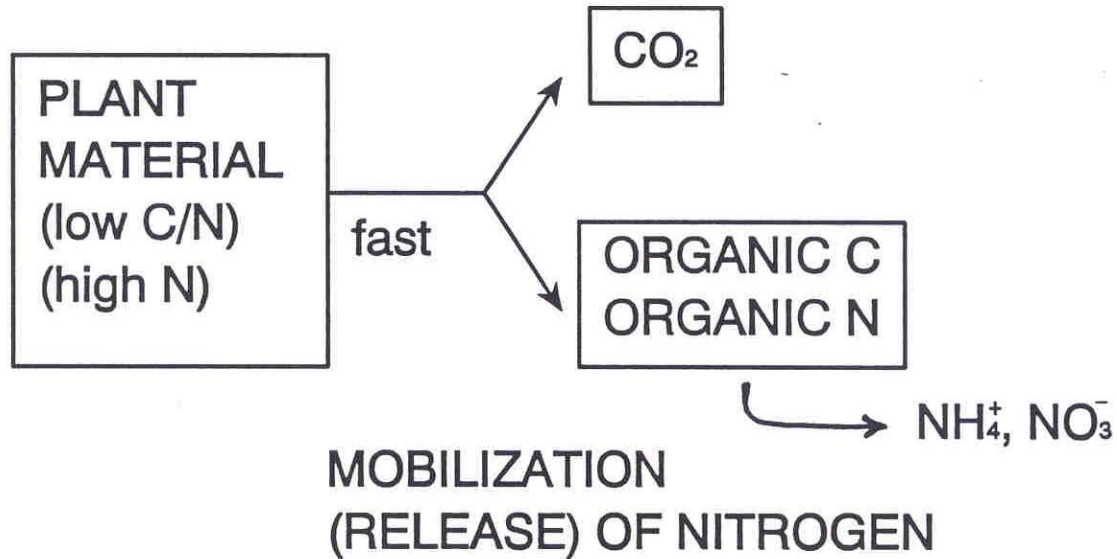
### **Materials inducing N deficiency:**

**Sawdust, wood chips, straw (not peat moss because of its slow decay)**

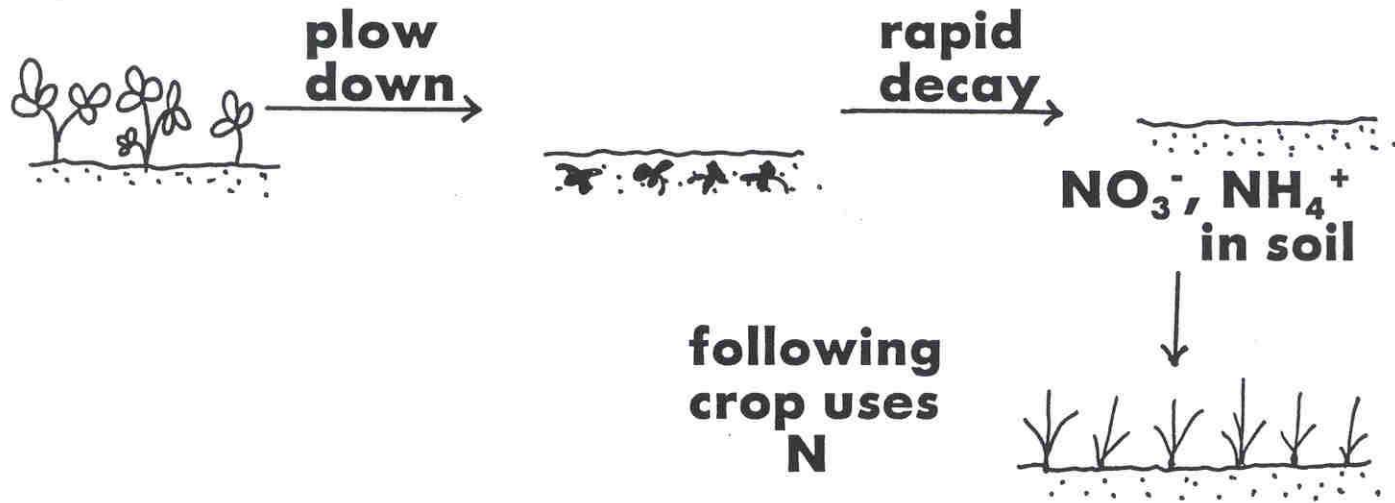


## Nutrient Release and Immobilization

$C/N < 20$  ( $\% N > 2.5$ )  $\rightarrow$  N release

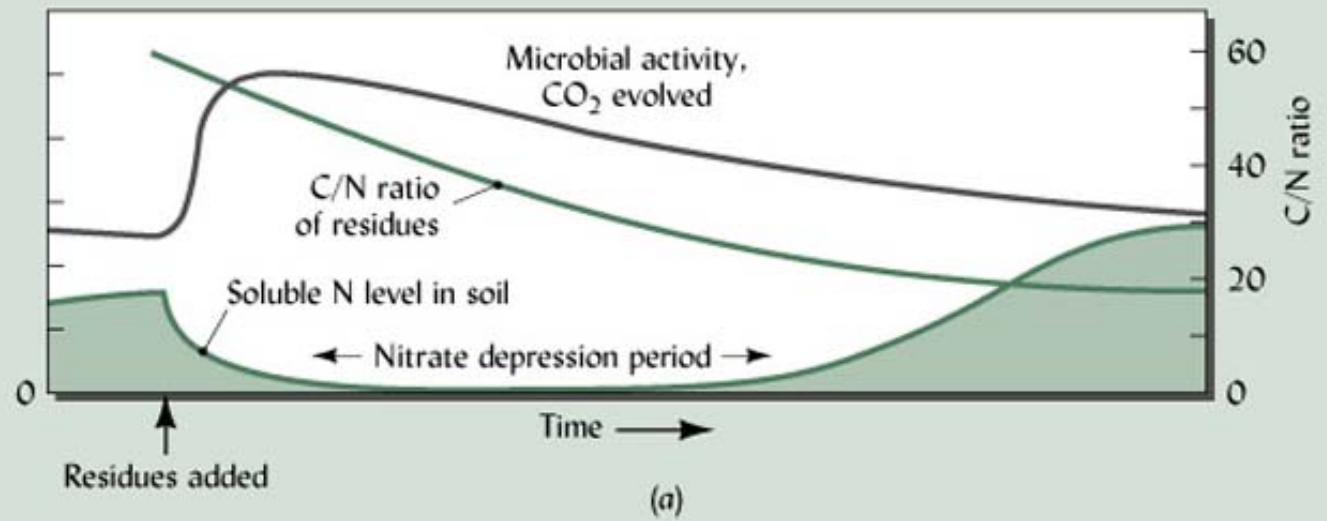


## Example: Clover (low C/N ratio)

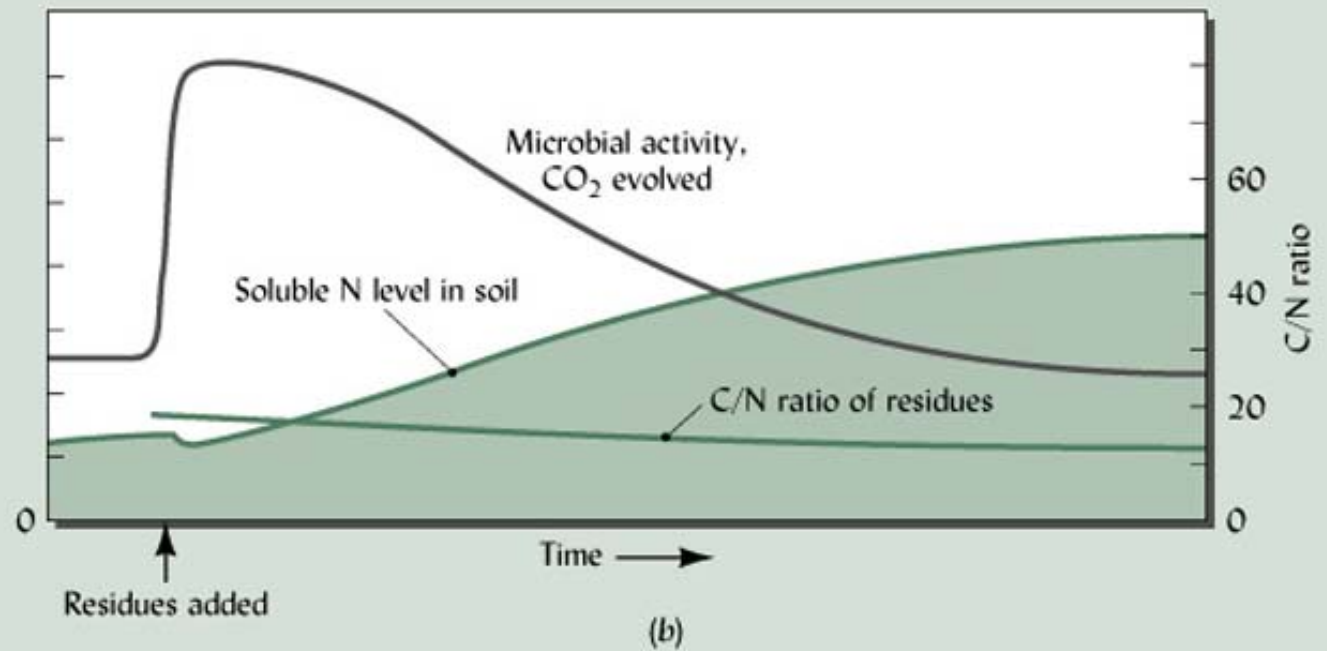


**Materials inducing N release:** legumes  
(green manures)

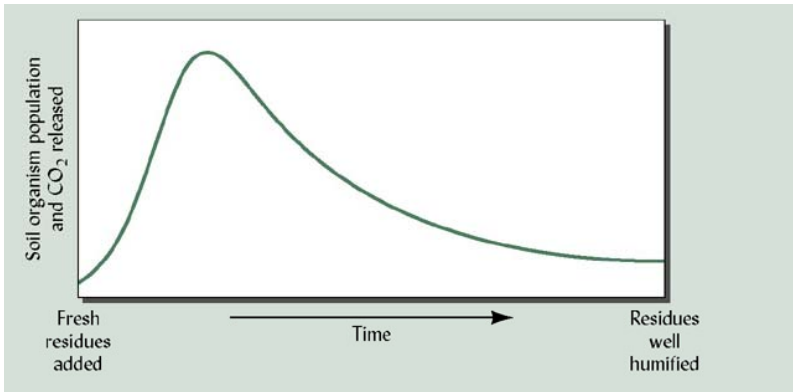
C:N ↑



C:N ↓



# Addendum: Influence on Soil Ecology



Something preys on the large decomposer population

